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Every Good Work

# SACRED HEART COLLEGE (AUTONOMOUS)

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A Don Bosco Institution of Higher Education, Founded in 1951 \* Affiliated to Thiruvalluvar University, Vellore \* Autonomous since 1987

Accredited by NAAC (4<sup>th</sup> Cycle – under RAF) with CGPA of 3.31 / 4 at 'A+' Grade

## Name of the Programme: B Sc. Chemistry

S No	Title of the Paper	Course Code	Course Objectives	Course Outcomes	Relevance
1	ORGANIC CHEMISTRY – I	CH116	<ul style="list-style-type: none"><li>Understanding the fundamentals of acidity and basicity.</li><li>Understanding the structure of organic compounds.</li><li>Fathom the acidity and basicity of organic molecules.</li><li>Provide the rudimentaries of stereochemistry.</li><li>Conceptualize the correlation between structure, acidity and reactivity.</li><li>Discern the reactivity of alkenes and alkynes.</li></ul>	<ul style="list-style-type: none"><li>On successful completion of this Course, students will be able to</li><li>Network structure, hybridization, acidity, basicity and reactivity of organic molecules, identify the molecular structure.</li><li>Categorize molecules on the basis of hybridization; predict the acidity and basicity of the molecules based on functional groups; predict products for organic reactions.</li><li>Represent molecules dimensionally; infer the reactivity of molecules from their hybridization; Chart out the mechanism for organic reactions.</li><li>Distinguish the stereochemistry of molecules; rationalize the reactivities of alkenes and alkynes.</li></ul>	Local developmental needs

				<ul style="list-style-type: none"> <li>• Evaluate and hypothesize the stability of molecular intermediates and alicyclic molecules.</li> <li>• Collaboratively assess the reactivity of molecules based on structure, stereochemistry, hybridization, acidity and basicity; Build on understanding the reactivity and mechanism of molecules and reagents.</li> </ul>	
2	ALLIED CHEMISTRY - I (BIOCHEMISTRY)	ACH110	<ul style="list-style-type: none"> <li>• To understand the basics of solution chemistry</li> <li>• To categorize the organic compounds based on aromaticity, hybridization, bonding</li> <li>• To integrate the concept of organic molecules chemistry in biological systems</li> <li>• To tabulate acidity and basicity concepts based on various theories</li> <li>• To reflect chemical kinetics concept through biological systems</li> <li>• To build relationship between the drug actions with chemical structure</li> </ul>	<ul style="list-style-type: none"> <li>• On successful completion of this Course, students will be able to</li> <li>• Understand the basics of solution chemistry and knowledge of preparing solutions with different concentrations</li> <li>• Categorise the organic compounds based on aromaticity, hybridization, types of bonding</li> <li>• Integrate the concept of buffer solution in chemistry with biological systems</li> <li>• Tabulate acidity and basicity concepts based on various acid-base theories</li> <li>• Reflect the concept of chemical kinetics in biological systems</li> <li>• Develop relationship and synthetic methodology of drugs such as anesthetics and antibiotics and establish the mode of action with biological systems.</li> </ul>	Local developmental needs

3	INORGANIC CHEMISTRY – I	CH216	<ul style="list-style-type: none"> <li>• To understand the basic atomic structure of elements their periodic properties and chemical bonding.</li> <li>• To evaluate the nature of bonding by applying various fundamental theories.</li> <li>• To learn the properties and applications of s and p block elements.</li> <li>• To compare and contrast the relationship between groups.</li> <li>• To understand the principles and theories of Acids and Bases.</li> <li>• To apply fundamental theories of acids and bases and identify the progress of the chemical reaction.</li> </ul>	<ul style="list-style-type: none"> <li>• On successful completion of this Course, students will be able to</li> <li>• Understand the behaviour and properties of the elements in the periodic table and comprehend them</li> <li>• Comprehend the fundamentals of electronic configuration, oxidation states, and specific properties of the major group elements.</li> <li>• Compare and contrast the properties of acids and bases and justify their applications</li> <li>• Predict atomic structure, chemical bonding, Hybridization, and molecular geometry.</li> <li>• Analyze and understand the diagonal relationship between alkali and alkaline earth metals their properties and applications.</li> <li>• Evaluate the properties of elements based on their atomic structure, bonding nature, etc., and relate the uses and significance of the s and p block elements.</li> <li>• Devise and validate acid and base using the metal oxides and predict the feasibility of the reaction</li> </ul>	Local developmental needs
4	CHEMISTRY IN EVERY DAY LIFE	NCH604	<ul style="list-style-type: none"> <li>• To appreciate the importance of chemistry in day-to-day appliances</li> <li>• To understand the role of chemicals involved in</li> </ul>	<ul style="list-style-type: none"> <li>• List the different types of soil and classify the fertilizers</li> <li>• Recognize the different household chemicals used and analyse their adverse effects</li> </ul>	Local developmental needs

			different substances, their uses and precautions to be taken.	<ul style="list-style-type: none"><li>• Tabulate the variety of glasses and ceramics available and apply them for specific purpose intended</li><li>• Identify the various types of adulterants present in food and demonstrate to identify them</li><li>• Evaluate the chemicals used in various cosmetic items and compare them</li><li>• Identify the components present in different plastics and hypothesize its uses</li></ul>	
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